









Universidad de Oviedo Universidá d'Uviéu University of Oviedo





# FUNMIN

## **FUN**damental Studies of **MIN**eral Carbonation with Application to CO<sub>2</sub> Utilisation

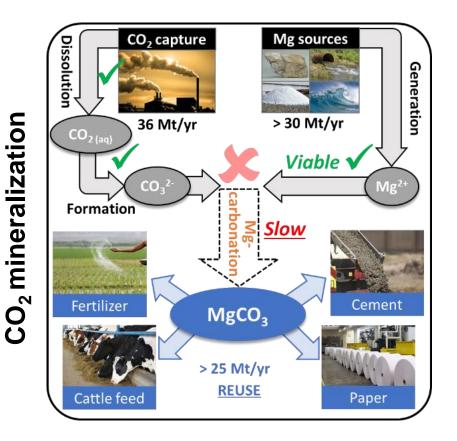


By Devis Di Tommaso, Queen Mary Presented at ACT workshop 07.11.2019



## CO<sub>2</sub> into added-value products

"CCUS can create new industries and markets through the use of carbon dioxide, such as chemicals, plastics, and building materials" \*



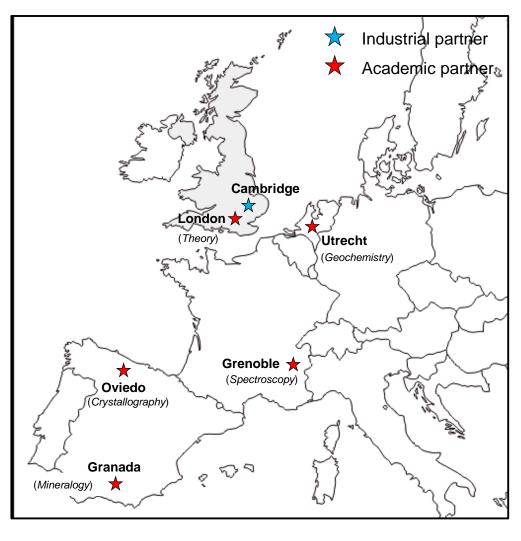


Cambridge Carbon Capture Ltd technology (**CO2LC**) to store CO<sub>2</sub> in mineral form (MgCO<sub>3</sub>)

# The FUNMIN consortium

## World expertise in mineralization guiding Industrial technologists to permanently mineralise CO<sub>2</sub>

$$CO_{2 \text{ (gas)}} \rightarrow MgCO_{3 \text{ (solid)}}$$

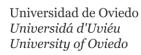
















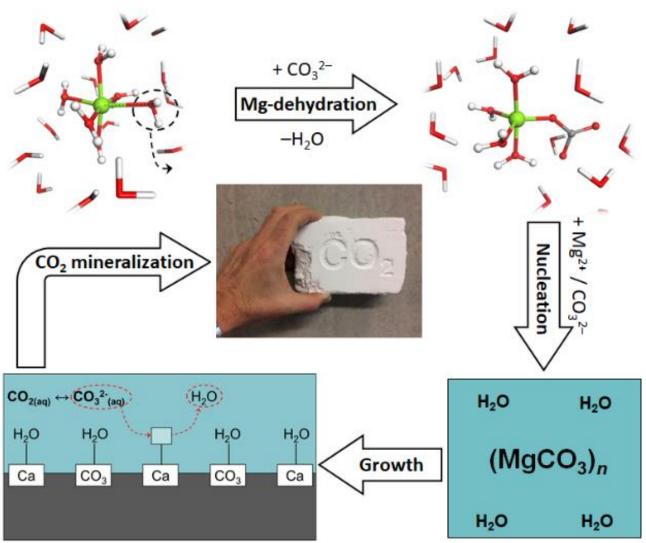


## FUNMIN facts

- Full Economic Cost: € 890k (€ 700k from ACT + € 190k in-kind)
- Duration: **30 months** (10/2019 03/2022)
- Academic partners: QMUL (coordinator), UGR, UO, UGA, UU
- Industrial partner: Cambridge Carbon Capture Ltd
- Associate partners: National Physical Laboratories (UK), McMaster University (Canada), University of Hong Kong (China), Seoul National University (Korea)
- ISIS Neutron and Muon Source facility at Rutherford Appleton Laboratory (UK)

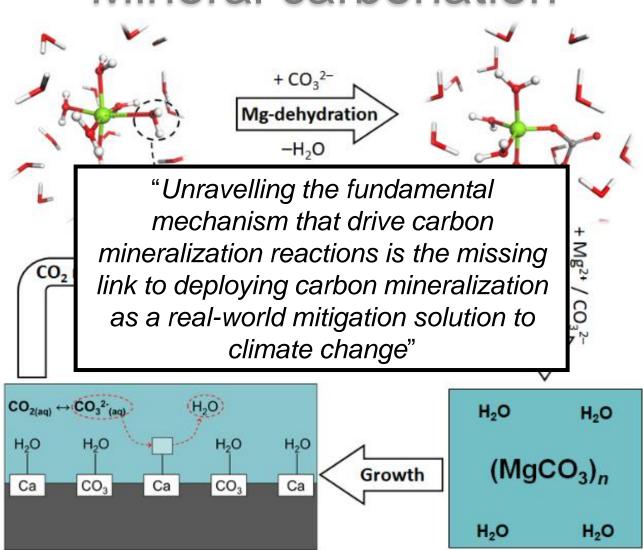


## **Mineral carbonation**





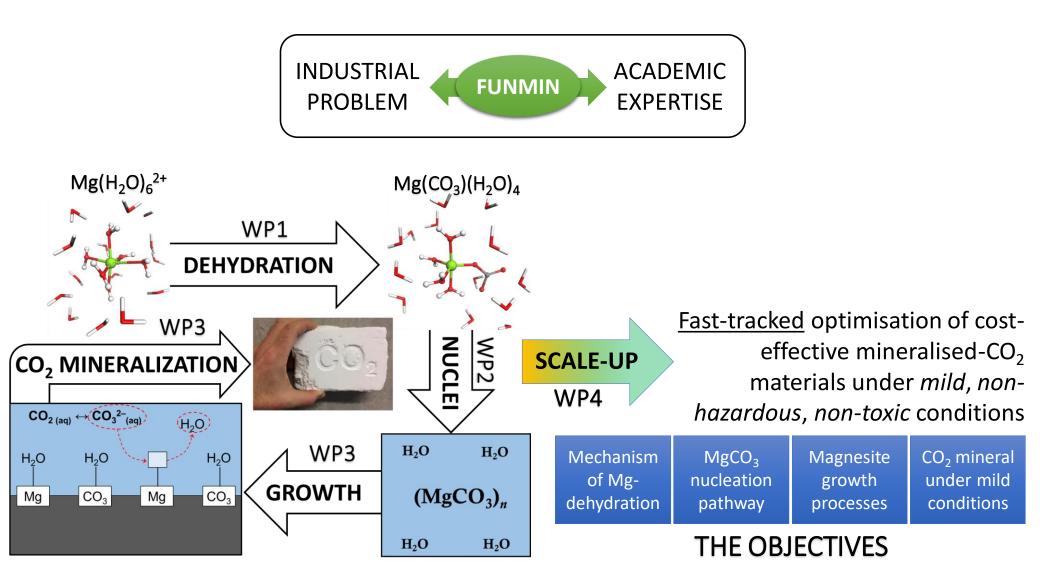




<u>CO<sub>2</sub> Utilization Priority Research Direction U5</u>: Accelerating Carbon Mineralization by Harnessing the Complexity of Solid-Liquid Interfaces, in "Mission Innovation Carbon Capture, Utilization, and Storage Workshop"



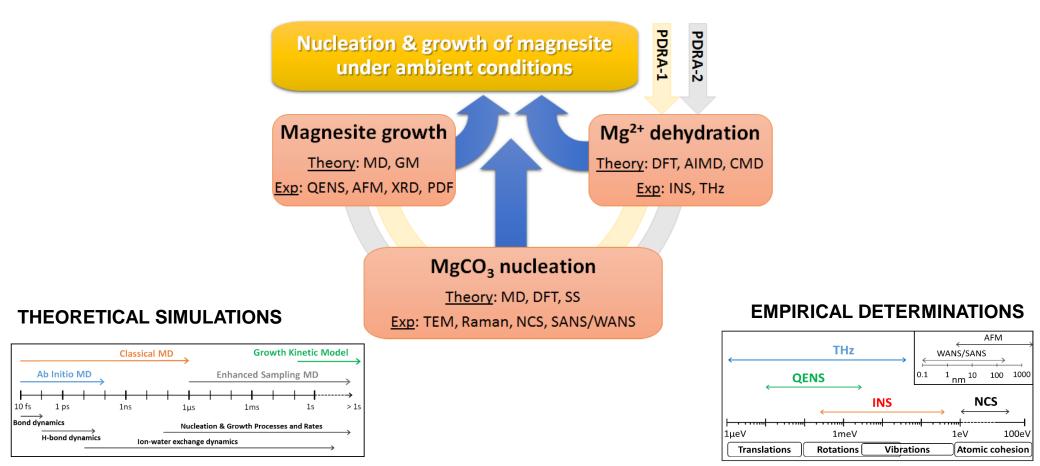
## The objectives of FUNMIN





## The FUNMIN approach

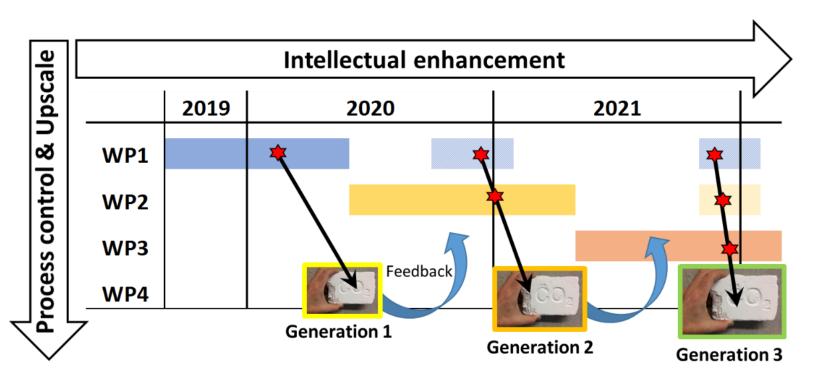
Complementary **atomistic simulations** & **spectroscopic measurements** to reveal the molecular-level processes controlling MgCO<sub>3</sub> formation: Mg-dehydration, MgCO<sub>3</sub> nucleation & growth.





## **Project implementation**

Core scientific activities (**WP1-3**) to characterize the **molecular processes** controlling magnesite crystallization; applied component (**WP4**) to **optimise** process conditions



Interaction between the scientific (**WP1-3**) and practical (**WP4**) components of the project. **WP1**: Mg-dehydration; **WP2**: MgCO<sub>3</sub> nucleation; **WP3**: Magnesite Growth; **WP4**: Upscaling



## **Project implementation**

WP = Working Package																SIM			c													
T = Task							1									SIIVI	ULAI		3												1	
D = Deliverables																EXPE	ERIN	IENT	S													
M = Selected milestone indicating key deliverables																																
			1		3	4	5	6	7	8	9	-		12	13	14	15	16	17	18	19	20	21	-		24	25	26	27	28	29	30
	Location	Activity		20	19							202	20											20	21						202	22
WP1 Mg-Dehydration			S	0	Ν	D	J	F	Μ	Α	м	J	J	Α	S	0	Ν	D	J	F	Μ	Α	Μ	J	J	Α	S	0	Ν	D	J	F
T1.1.1 Mechanism: DFT, AIMD and AIMμD calculations	QMUL	SIM				D1.1		_																								
T1.2.1 Assessment & calibration of forcefields	QMUL	SIM					D1.2																									
T1.3.1 Kinetics of Mg-dehydration in elect solutions	QMUL	SIM								D1.3																						
T1.4.1 AIMD of VDOS of Mg2+ in elec solutions	QMUL	SIM																														
T1.4.2 THz expt of (sub)ps H2O dynamics	QMUL/NPL	EXP									D1.4																					
T1.4.3 INS expt of the low-frequency H2O dynamics	UGA	EXP					M1																									
WP2 MgCO3 nucleation			S	0	Ν	D	J	F	м	Α	М	J	J	Α	S	0	Ν	D	J	F	М	Α	м	J	J	Α	S	0	Ν	D	J	F
T2.1.1 Mechanism: DFT and AIMD calculations	QMUL	SIM														D2.1																
T2.2.1 Structure & stability of MgCO3 nuclei	QMUL	SIM																				D2.2										
T2.3.1 Chemical analysis of hydrated/anhydrous PNC	UGA	EXP																	D2.3													
T2.3.2 Size & morphological PNC analysis	UGR	EXP																	UZ.5													
T2.4.1 Tracking particle formation (NCS, WANS/SANS )	QMUL/RAL	EXP														N	14					D2.4										
WP3 MgCO3 growth			S	0	Ν	D	J	F	М	Α	м	J	J	Α	S	0	Ν	D	J	F	м	Α	м	J	J	Α	S	0	Ν	D	J	F
T3.1.1 Solid state Magnesite crystal characterization	UO	EXP																														
T3.1.2 X-ray PDF of anhydrous and hydrated MgCO3	UO	EXP									М3																03.1					
T3.1.3 Solid-state DFT calculations magnesite crystals	UO	SIM																														
T3.2.1 MD of struct heterogeneous MgCO3 surfaces	QMUL/UU	SIM																														
T3.2.2 AFM experiments of magnesite growth	UGR	EXP																									Μ	<b>6</b>		C	)3 <mark>.</mark> 2	
T3.2.3 QENS experiments of surface H2O dynamics	QMUL/RAL	EXP																														
T3.3.1 Development of surface complexation model	UU	SIM																												C	03.3	
T3.4.1 Development of growth kinetic model	UU	SIM																			M5										C I	D3.4
WP4 Practical upscale			S	0	Ν	D	J	F	м	Α	м	J	J	Α	S	0	Ν	D	J	F	М	Α	м	J	J	Α	S	0	Ν	D	J	F
T4.1.1 In-situ Raman analysis of nucleating particles	UGA	EXP										<b>D4.1</b>																				
T4.2.1 Characterisation of sample batches during mineral	QMUL/RAL	EXP							<mark>-</mark> -													D4.2										
T4.2.2 Formulating of metrics	QMUL/RAL	EXP																				D4.2									M	•
T4.2.3 NCS, SEM and TEM imaging of samples	QMUL/RAL	EXP																														
T4.3.1 Industrial demonstration rig experiments	CCC/QMUL	EXP																	Ŀ				-								ľ	D4.3
WP5 Management & dissemination			S	0	Ν	D	J	F	М	Α	м	J	J	Α	S	0	Ν	D	J	F	м	Α	м	J	J	Α	S	0	Ν	D	J	F
T5.9 Organise project's meetings	1			UK																ES								_				FR



## **Project implementation**

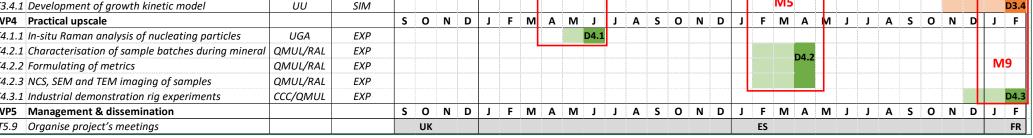
W/P = \	Norking Package		_	No.
T = Tas				
	liverables			M1
M = Se	elected milestone indicating key deliverables			
				<b>M2</b>
		Location	Act	
WP1	Mg-Dehydration			
T1.1.1	Mechanism: DFT, AIMD and AIMµD calculations	QMUL	Si	MO
T1.2.1	Assessment & calibration of forcefields	QMUL	SI	M3
T1.3.1	Kinetics of Mg-dehydration in elect solutions	QMUL	SI	
T1.4.1	AIMD of VDOS of Mg2+ in elec solutions	QMUL	SI	
T1.4.2	THz expt of (sub)ps H2O dynamics	QMUL/NPL	E.	Μ4
T1.4.3	INS expt of the low-frequency H2O dynamics	UGA	E.	101-
WP2	MgCO3 nucleation			М5
T2.1.1	Mechanism: DFT and AIMD calculations	QMUL	SI	IVIJ
T2.2.1	Structure & stability of MgCO3 nuclei	QMUL	SI	M6
T2.3.1	Chemical analysis of hydrated/anhydrous PNC	UGA	E.	OIVI
T2.3.2	Size & morphological PNC analysis	UGR	E.	
T2.4.1	Tracking particle formation (NCS, WANS/SANS )	QMUL/RAL	E.	
WP3	MgCO3 growth			M7
T3.1.1	Solid state Magnesite crystal characterization	UO	E.	
T3.1.2	X-ray PDF of anhydrous and hydrated MgCO3	UO	E. E. Si	<b>M8</b>
T3.1.3	Solid-state DFT calculations magnesite crystals	UO	S	IVIO
T3.2.1	MD of struct heterogeneous MgCO3 surfaces	QMUL/UU	Si E.	N/O
ТЗ.2.2	AFM experiments of magnesite growth	UGR	E.	M9
T3.2.3	QENS experiments of surface H2O dynamics	QMUL/RAL	E.	
T3.3.1	Development of surface complexation model	UU	E. Si	
T3.4.1	Development of growth kinetic model	UU	SII	И
WP4	Practical upscale			:
T4.1.1	In-situ Raman analysis of nucleating particles	UGA	ΕX	(P
T4.2.1	Characterisation of sample batches during mineral	QMUL/RAL	ΕX	(P
T4.2.2	Formulating of metrics	QMUL/RAL	ΕX	(P
T4.2.3	NCS, SEM and TEM imaging of samples	QMUL/RAL	ΕX	(P
T4.3.1	Industrial demonstration rig experiments	CCC/QMUL	EX	(P
WP5	Management & dissemination			
T5.9	Organise project's meetings			

#### -

- Assessment & calibration of atomistic models
- Private Pri

**Milestone Title** 

- Effect of additives promoting Mg-dehydration, on the kinetics of crystallization of anhydrous and hydrated Mg-carbonate phases
- Theoretical model of MgCO<sub>3</sub> nucleation from solution developed
- In-situ tracking of mechanical properties & changes therein
- Mechanistic model of the magnesite growth in aqueous electrolyte solutions
- Site-specific surface complexation model for calcite developed
- Site-specific growth kinetic model for magnesite
- Factors catalysing magnesite crystallization under mild, nonhazardous, non-toxic conditions identified





## FUNMIN Outreach plan

Dissemination audience	Dissemination Goal	Methods of Dissemination						
Other Researchers working on CCUS	Understanding	Presentations and Professional Networking; Academic journals; Social media; Project's website.						
Policy makers	Awareness, Understanding, Action	Formal Reports; Project meetings; Sharing research outputs.						
Future Funding Organisations	Awareness, Understanding, Action	Formal reports; Social media; Sharing research outputs.						
The Wider Community	Awareness	Social media; Project's website; Sharing research outputs; Outreach.						

Objectives of our dissemination strategy :

- Inform other researchers (Chemists, Geoscientists, Industrial scientists) working on CCUS or related fields
- **Bridge** the gap between research and policy
- Shape funding strategy of UK and EU research councils towards CO<sub>2</sub> mineralization technologies
- Interact with the public



## FUNMIN Outreach plan

### Presentation & professional networking

- Presented at the FIRED-Up event (06.11.2019)
- Press release on FUNMIN with <u>www.consciouscomms.com</u>
- Kick-off meeting in London (13.12.2019)
- Canada-UK Communities of Interest: Commercial CO<sub>2</sub> Capture and Use Opportunities (26.11.2019)
- Conferences and events organised by UKCCSRC, BACG, TYC and NMUSN

### Website & Social media

http://research.sbcs.qmul.ac.uk/d.ditommaso/funmin



### • Sharing research outputs

Conference papers, software, posters, presentations, reports, protocols, preprint on FUNMIN website



# FUNMIN contribution to commercialisation of $CO_2$ mineralization

### Scientific challenge

 What are the fundamental processes optimising the aqueous formation of MgCO<sub>3</sub> at low temperature in nature?

### Technical challenge

– What are the process conditions that could catalyse magnesite formation under mild conditions?

### Commercial challenge

 Can we develop cost-effective processes for the selective conversion of CO<sub>2</sub> into magnesite under mild, non-hazardous, and non-toxic conditions?

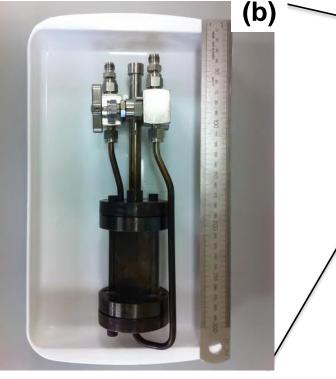
### Provide (theory & expt) tools to the CO<sub>2</sub> mineralization industry

- To raise its impact and competitiveness



## Interface between Cambridge Carbon Capture existing carbonation rig and the neutron beam facility





Special Nickel-Chromium alloy for high pressure conditions

- a) CCC's experimental set-up to observe  $CO_2$  reaction with  $Mg(OH)_2$
- b) Stainless steel reaction cell for neutron measurements of heterogeneous catalyst samples (Johnson Matthey Technol. Rev., 2016, 60, 132)

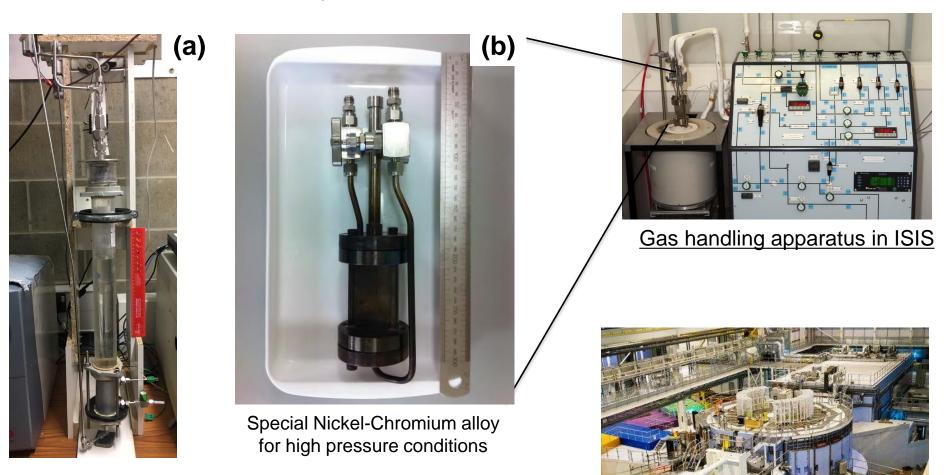
#### Gas handling apparatus in ISIS



Rutherford Appleton Laboratory (UK)



## Interface between Cambridge Carbon Capture existing carbonation rig and the neutron beam facility



- a) CCC's experimental set-up to observe CO<sub>2</sub> reaction with Mg(OH)<sub>2</sub>
- b) Stainless steel reaction cell for neutron measurements of heterogeneous catalyst samples (Johnson Matthey Technol. Rev., 2016, 60, 132)

Rutherford Appleton Laboratory (UK)



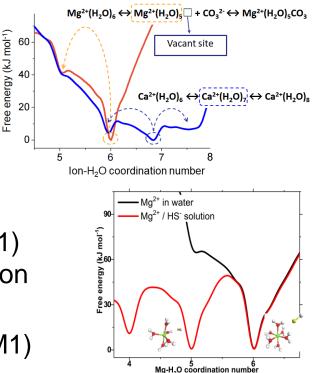
# Status of FUNMIN project

### Contracts and Consortium agreement

- UK and France contracts in place. Consortium agreement signed. Spanish contract will be in place early 2020
- QMUL hired Computational Chemist (PDRA-1) and Physical Chemist (PDRA-2)

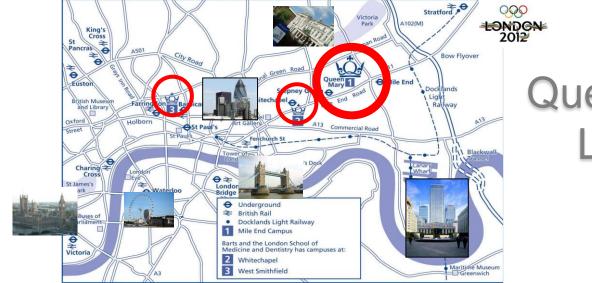
### First results

- Proposals for neutron experiments submitted to ISIS
  Neutron and Muon Source (UK)
- Awarded IAA Grant to develop a flow-cell for neutron scattering measurements of CO<sub>2</sub> mineralisation
- Water exchange reaction pathways around Mg<sup>2+</sup> (D1.1)
  Dynamics of water around Mg<sup>2+</sup> as a function of solution composition (D1.3)
- Assessment & calibration of interatomic force fields (M1)





## Contact us



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### **Devis Di Tommaso**

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## Acknowledgements



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http://research.sbcs.qmul.ac.uk/d.ditommaso/funmin